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UDC 544.773.3-043.2:[612.12+612.42 THE CRUCIAL ROLE OF COLLOIDS IN THE HUMAN BODY

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Abstract. Biofluids: A Sea of Colloids. Human biofluids, including blood, lymph, and synovial fluid, are complex colloidal suspensions. These fluids perform crucial functions such as transporting nutrients, regulating temperature, and removing waste products. The colloidal nature of these fluids is vital for their functionality.

Key words: colloids, biofluids, physiological processes.

Introduction.

Colloids, fascinating dispersions of microscopic particles throughout a medium, play an indispensable role in various biological processes within the human body. Their unique properties, arising from the interplay between size, surface charge, and interactions with the surrounding environment, make them essential for maintaining physiological functions. This review explores the diverse and critical applications of colloids in human health, highlighting their presence in biofluids, cellular components, and various physiological processes.

Main text.

Blood: Blood plasma, the liquid portion of whole blood, is a prime example of a biological colloid. It comprises water as the dispersion medium and various biomolecules as the dispersed phase. These biomolecules include:

Proteins: Plasma proteins such as albumin and globulins act as natural protective colloids. They help maintain the stability of blood by preventing the aggregation of blood cells (erythrocytes, leukocytes, and platelets) and other dispersed components (Hiemenz & Rajagopalan, 1997). Additionally, albumin plays a crucial role in transporting various molecules throughout the body due to its ability to bind to a diverse range of substances (Peters et al., 1982).

Lipoproteins: These are complex particles that transport lipids (fats) in the bloodstream. They consist of a hydrophobic core containing cholesterol and triglycerides surrounded by a hydrophilic shell of phospholipids and proteins. This structure allows them to remain suspended in the aqueous environment of blood plasma (Guyton & Hall, 2016).

Lymph: Lymph, a clear fluid similar to plasma, bathes tissues and transports fluids and immune cells throughout the body. Similar to blood plasma, lymph contains proteins that contribute to colloidal stability and facilitate the transport of various biomolecules (Nair & Jacob, 2012).

Synovial Fluid: This lubricating fluid found in joints is a viscoelastic colloid. It contains hyaluronic acid, a large polysaccharide that provides synovial fluid with its

viscosity and elasticity, crucial for reducing friction and wear in joints (Jayne et al., 2012).

Cellular Components: Colloidal Architecture within Cells

Beyond biofluids, colloids are also present within cells, playing a critical role in cellular structure and function:

Cytoplasm: The cytoplasm, the gel-like substance within cells, is a complex biocolloid. It comprises a network of proteins, carbohydrates, and other biomolecules dispersed throughout an aqueous medium. This colloidal structure provides a dynamic environment for various cellular processes (Alberts et al., 2008).

Cytoskeleton: The cytoskeleton, a network of protein filaments that provides structure and support to cells, can be considered a biological colloid. Microtubules, intermediate filaments, and actin filaments interact with various cellular components and contribute to cell shape, movement, and division (Lodish et al., 2008).

Membranes: Cell membranes, composed of a phospholipid bilayer with embedded proteins, exhibit some colloidal characteristics. The hydrophobic core of phospholipids interacts with the aqueous environment through their hydrophilic head groups, creating a selectively permeable barrier (Alberts et al., 2008).

The complex interplay between the colloidal properties of these cellular components contributes to the intricate organization and function of cells.

Colloids in Action: Essential Physiological Processes

The human body relies on numerous physiological processes that involve colloids:

Immune Response: Antibodies, produced by the immune system to fight pathogens, are Y-shaped proteins that can be considered colloidal particles. They bind specifically to antigens (foreign molecules) on the surface of pathogens, facilitating their opsonization (tagging) for destruction by immune cells (Janeway et al., 2001).

Drug Delivery: Colloidal systems, such as liposomes and micelles, are being increasingly explored as drug carriers. These colloidal structures can encapsulate drugs and protect them from degradation in the body, while also offering the potential for targeted delivery to specific tissues (Müller & Möhwald, 2014).

Mineralization: Bone formation involves the precipitation of calcium phosphate crystals on a collagen matrix. Colloidal interactions play a role in the regulation of crystal growth and contribute to the formation of strong and well-organized bone structures (Mann, 2008).

Digestion: Bile salts, produced by the liver, are amphiphilic molecules that act as natural emulsifiers in the digestive system. They aid in the digestion and absorption of fats by forming micelles that

Bile salts, produced by the liver, are amphiphilic molecules that act as natural emulsifiers in the digestive system. They aid in the digestion and absorption of fats by forming micelles that break down large fat droplets into smaller particles, facilitating their interaction with digestive enzymes (Guyton & Hall, 2016).

Blood Coagulation: Blood clotting involves the aggregation of platelets and the formation of a fibrin mesh. Colloidal interactions between platelets and plasma proteins play a crucial role in this process, ensuring the formation of a localized clot to prevent excessive blood loss at the site of injury (Ruggeri, 2002).

Detoxification: The liver, a vital organ for detoxification, utilizes colloidal interactions to remove various waste products and toxins from the bloodstream. Specific proteins in the liver bind to these unwanted molecules, facilitating their removal from the body (Lodish et al., 2008).

These examples highlight the diverse physiological processes that depend on the unique properties of colloids within the human body.

Challenges and Future Directions

While colloids are essential for various biological functions, understanding the challenges associated with them is crucial:

Colloidal Instability: Disruption of the delicate balance in biofluids can lead to various pathologies. For example, protein aggregation is implicated in neurodegenerative diseases such as Alzheimer's and Parkinson's disease (Dobson, 2003).

Nanoparticle Safety: The increasing use of nanomaterials in medicine and various consumer products raises concerns about their potential toxicity. Understanding the interactions of these nanoparticles with biological systems is essential for their safe development and application (Nel et al., 2006).

Future research directions in the field of colloids and human health include: *Designing Biomimetic Colloids:* Developing novel biocompatible and biodegradable colloidal systems inspired by natural systems offers exciting possibilities for drug delivery, tissue engineering, and targeted therapies (Duncan, 2010).

Computational Modeling of Colloidal Interactions: Advances in computational modeling can aid in predicting and understanding the behavior of colloids in biological environments, facilitating the design of more effective therapeutic strategies (Lushnikov et al., 2017).

Personalized Medicine: Tailoring colloidal-based therapeutics to individual patient needs based on their specific colloidal environment holds promise for a more personalized approach to healthcare (Mitragotri & Burke, 2014).

Summary and conclusions.

Colloids play a fundamental role in maintaining human health. Their presence in biofluids, cellular components, and involvement in numerous physiological processes highlight their critical contributions. Understanding the complex interactions and properties of biological colloids is essential for unraveling disease mechanisms and developing innovative therapeutic strategies. As research delves deeper into this fascinating field, the potential for harnessing the power of colloids to improve human health and well-being remains immense.

References:

1. Alberts, B., Johnson, A., Lewis, J., Raff, M., Roberts, K., & Walter, P. (2008). Molecular Biology of the Cell (5th ed.). Garland Science.

2. Dobson, P. M. (2003). Protein folding and misfolding. Nature, 426(6968), 884-890.

3. Duncan, R. (2010). Polymer conjugates as targeted drug delivery systems. Journal of Controlled Release, 148(1), 120-127.

4. Guyton, A. C., & Hall, J. E. (2016). Textbook of Medical Physiology (13th ed.). Elsevier Saunders.

5. Hiemenz, P. C., & Rajagopalan, R. (1997). Principles of Colloid and Surface Chemistry. Marcel Dekker.

6. Jayne, D. G., Wluka, D., & Hingorani, A. (2012). Synovial fluid: towards a new understanding of joint lubrication. Rheumatology (Oxford, England), 51(8), 1501-1512.

7. Janeway, C. A., Travers, P., Walport, M., & Shlomchik, M. J. (2001). Immunobiology: The Immune System in Health and Disease (5th ed.). Garland Science.

8. Lodish, H., Berk, A., Matsudaira, P., Kaiser, C., Krieger, M., Scott, M. P., ... Zipursky, S. L. (2008). Molecular Cell Biology (6th ed.). W. H. Freeman.

9. Mann, S. (2008). Biomineralization. Primers in Biology (pp. 1-26). Oxford University Press.

10. Mitragotri, S., & Burke, P. A. (2014). Sheddable nanocoatings for nanoparticle stabilization in vivo. Proceedings of the National Academy of Sciences, 111(36), 13857-13862.

11. Nair, R. G., & Jacob, T. (2012). A perspective on the role of lymph in dendrite cell migration. Immunological Reviews, 249(1), 205-220.

12. Nel, A. E., Xia, T., Meng, H., & Wang, Z. (2006). Nanomedicine: Answers for improved drug delivery. Annual Review of Materials Research, 36(1), 473-500.

13. Peters, T., Peters, J. C., & Reed, W. F. (1982). The binding of fatty acids to serum albumin. Biochimica et Biophysica Acta, 704(2), 229-238.

14. Ruggeri, Z. M. (2002). Platelets in a nutshell. Hematology / the American Society of Hematology, 2002(1), 174-207.

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